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## ASSESSMENT AND FORECAST FOR THE CREATION OF PHOTOCHEMICAL SMOG OVER TRANSPORT OVERPASSES IN KYIV

***Abstract.** This article deals with the analysis of existing models of smog situation formation over automobile overpasses and in places of substantial congestion of transport in large cities, for example, in Kyiv. A mathematical model consisting of two blocks – dynamic and kinetic, which allows determining the formation of the thermal dome of pollution and the concentration of hydrocarbon emissions in the air, depending on the number of working engines, is proposed. The kinetic block of the model allows determining the level of formaldehyde, as an indicator of the appearance of photochemical smog in conditions of atmospheric constancy. The concentration of emissions from motor vehicles in the air is calculated over the main transport overpasses of the city of Kyiv at their full load (peak hours).*

***Key words:** bridge, smog, formaldehyde, indicator of pollution, formation of a dome of pollution, concentration of pollution, hydrocarbons, rate of conversion, mathematical model of atmospheric pollution.*

### Introduction

The main factors and mechanism of formation of smog in large cities of Ukraine, including over the city of Kyiv, studied to date is not enough. Most works are devoted to a comprehensive assessment of the level of air pollution, where one of the leading contributing factors is formaldehyde. Formaldehyde proliferation in large cities is being addressed by such domestic and foreign scientists as Polischuk S.Z., Dotsenko L.V., Bezuglaya E.Y., Snizhko S.I., Shevchenko O.G., Kulbida M.I., Lezhenin A.O., Vorobyova I.A., H. Bridgman, H. John Seinfeld, L. Graham, and others. Recently, formaldehyde pollution of the urban environment and simulation at the local level of this phenomenon is widely studied on the example of cities of the post-Soviet Union countries (Sellegy T.S., Shlychkov V.A., Malbakhov V.M., Lenkovskaya T.N., Temirbekov and others), where there is a rather significant load on the atmosphere, in particular from point sources of pollution.

For the city of Kyiv average daily concentration of formaldehyde in summer can exceed twice the sanitary-and-hygienic standard, as evidenced by data from the Central Geophysical Laboratory of the Ukrainian Hydro Meteorological Center [5, 10, 18]. Many authors note that in large cities of Ukraine, and in Kiev, in the first

place, the concentration of formaldehyde can serve as a sign of the appearance of photochemical smog in urban accumulations of motor vehicles under steady weather conditions of the atmosphere.

The dynamics of the formation of smog is studied for a long time on an example of its creation in Los Angeles, London, Sydney and other major metropolitan areas of the world. According to the World Health Organization in urban airspace under adverse weather conditions or with intensive motor transport, the concentration of formaldehyde may reach  $0.1 \text{ mg} / \text{m}^3$  [13].

Although, the mechanism of the formation of smog, which was considered in the works of a number of foreign authors, is identical, the indicator of urban photochemical pollution of the atmosphere in each case is chosen different and depends on a number of factors of the area, primarily from the meteorological conditions of the terrain, the type of pollutants. For Kyiv, such an indicator can be the concentration of formaldehyde, as a product of oxidation of hydrocarbons from the emission of exhaust gases of various types of automobile engines and natural factors (hydrocarbons of vegetable origin). Taking into account the annual growth of motor transport on the roads of Kyiv and the correlation of the appearance of photochemical smog over automobile overpasses (especially on two or more levels) and on the traffic jams and the number of vehicles, the assessment of the formation of photochemical smog in these places is extremely important in order to provide a large city with sufficient level of environmental safety.

**The aim of research.** The work aims to reveal the peculiarities of the negative impact on the atmospheric air of motor transport in Kyiv, as a scientific basis for improving the environmental safety management system by estimating and predicting the formation of photochemical smog around overpasses and large intersections.

### **Presenting main material**

Among the existing models of the formation of photochemical smog over a large city at present, the most common are statistical-hydrodynamic, which are adapted to the conditions of a large city. As a predicted value, the integral index of air pollution by formaldehyde in the city was calculated, which was calculated using regression analysis methods. The statistical component of the model gives the average prognostic concentration in the city; the spatial distribution details are based on the hydrodynamic block of the model [6, 7, 11, 14, 15, 16, etc.]. The modeling of the distribution of harmful impurities in the atmosphere from point sources and vehicles, taking into account photochemical transformation, was considered in a number of studies using the model of the boundary layer of the atmosphere to take into account the influence of anthropogenic heat sources and the heterogeneity of the underlying surface on the process of spreading pollution [1, 6, 13, 14, 16]. In the works [2, 13, 14] a mechanism for the formation of photochemical smog at maximum concentrations of formaldehyde, that fixed in weak winds and calm, and a model for the creation of formaldehyde from ethane (emissions from motor vehicles) by the balance method are presented [2].

In this article the model of the formation of photochemical smog in the places of a significant accumulation of vehicles in Kyiv is discussed, which consists of two blocks, namely:

**The first block.** It is based on the assumption that a "dome" of warm air, like a heat source, forms over a transport overpass. The convective heat that gives the heat source to the environment determines the nature and parameters of the convective air jet that develops over the heated surface. At the first stage, the equation of the amount of motion of polluted air is solved, which allows determining the main parameters of the contaminated jet and calculating the amount of heat coming from the warm source into the environment, as well as the nature and parameters of the convective jet. In this case, we have an assumption regarding the circular area of the transport node S diameter D.

We find the amount of heat in the area S.

$$Q_s = S \cdot \sum Q_{av},$$

$\sum Q_{av}$  – the sum of direct and scattered radiation (for Kiev 645 + 194 = 893 MJ / m<sup>2</sup> respectively).

We find warmth from cars on the transport node.

By the number of lanes, we determine the number of vehicles N. In times of maximum load (peak times), we assume the interval between cars L = 20 m and 1 km of road in these times, there are 40 cars.

$$N = n \cdot L' / (l_{car} + L),$$

where  $l_{car}$  – a length of the car, m; n – the number of automobile lanes on the overpass, L – the length of one lane, m.

The heat from cars is according to the dependence:

$$Q_{car} = q_{car} \cdot N \cdot 40000 \text{ J},$$

where  $q_{car}$  – fuel consumption for one car per 1 m way, l.

According to the found value  $Q = Q_{car} + Q_8$ , we calculate the average temperature and velocity in the transverse section of the convective jet.

The average of the area of the warm air, which is rising upward and the average temperature in the transitional section, respectively, are based on the following formulas:

$$V_y = 0,56 (Q_K / (y - y_0))^{0,33},$$

$$\Delta t_{y_{cp}} = \frac{41 \cdot Q_K^{2/3}}{(y - y_0)^{5/3}}, \text{ Degree}$$

In the resulted formulas  $y - y_0$  – distance from the surface of the earth to the narrowest section of the thermal jet, m.

By source [17], the typical composition of exhaust gases (hydrocarbons CH<sub>1,85</sub>) by type of engine:

Type of Engine		
CE	Intermediate	$7,5 \frac{g}{kw \cdot h}$
DE	Load	$0,5 \frac{g}{kw \cdot h}$

When calculating the average amount of emissions from vehicles, we accept the average power of one car of 100 kilowatts and the smooth arrival and departure from the transport unit of cars at a speed of ~ 60 km / h.

Table 1 shows data on some overpasses in Kyiv, calculated according to the given methodology.

Concentration of emissions from car transport of hydrocarbons on the largest transport overpasses in Kyiv.

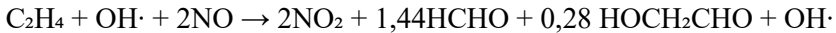
Table 1

Overpass	Area of thermal surface, m <sup>2</sup>	Heat from cars and warm surface at the transport node (per month), MJ / m <sup>2</sup>	Consumption in the narrowest section of the convective jet, m <sup>3</sup> / s	Concentration of emissions from vehicles at a distance of 2D above the surface of the overpass, mg / m <sup>3</sup>
1	2	3	4	5
Naddniprovske highway, st. Saperno-Slobidska (Vydubichi)	70650	63090450,28	70000	1,0
Avenue of the 40th anniversary of October, Nauky avenue (Central bus station)	96163	80680757	129513	0,4957
Mykoly Bazhana Ave, kil'tseva doroha	49086	41163857	73594,5	0,6200
Leningrad Square	61544	51635416	89222	0,5782
Industrial bridge, Borschagovska st.	96208	80718512	129574	0,4939
st. Saperno-Slobidska, street Kikvidze	70684	59303876	100165	0,5500
St. Akad. Zabolotnoho, Stolychne highway	125660	10548740	161944	0,4500
St. Bohatyr's'ka, st. Marshala Tymoshenko	61573	51635416	89222	0,5782

**The second block** provides for the consideration of kinetic models of converting the concentration of emissions of hydrocarbons from motor vehicles under certain weather conditions in formaldehyde.

Finding the concentration of ethane in a vertical convective stream at an altitude of up to 1 km above the surface of the overpass, we find the fraction of its transformation into formaldehyde under certain weather conditions.

The balance model of formation of formaldehyde from the inflow from mobile sources is presented in [2]:



As a result of the attack of hydroxyl radicals on ethylene at a temperature of 298 K, there is the conversion of two NO molecules into NO<sub>2</sub>, the formation of 1.44 formaldehyde molecules and 0.28 glycolic aldehyde molecules (HOCH<sub>2</sub>CHO) and regeneration of the hydroxyl radical. Comparing this mechanism with propane, we see the similarity of the conversion of NO to NO<sub>2</sub> and the formation of oxygen products.

The concentration of formaldehyde (secondary contamination) in the general form is written by the equation:

$$C_{\text{CHOH}} = K \cdot C_{\text{CH}},$$

where K is the coefficient that characterizes the reaction velocity or the conversion constant and depends on the temperature of the air, the time interval, the intensity of the sunlight, etc.

In works [1, 6, 14, 15] a method is presented for taking into account the transformation of harmful impurities in the atmosphere, taking into account photochemical transformations. For certain types of contaminants, the scheme of converting one substance to another under the action of temperature and humidity in the mixture is described by a system of differential equations, and each transformation has its rate constant, which is obtained by solving this system.

Another approach to determining the constant rate K is presented in [2, 4]. Gas molecules can react only when they are approaching each other for direct energy exchange, which can lead to the breakdown of connections. Since the collision of two molecules is a necessary condition for the reaction, there must be sufficient energy to break the chemical bond. This energy can be represented as  $\exp(-E/RT)$ , where R is a universal gas constant, T is temperature. The reaction rate is expressed in a form that defines both the collision frequency and the particle that exceeds the required energy for molecules of type i and j:

$$r = A(T) \cdot \exp(-E/RT) \cdot C^i \cdot C^j = K \cdot C^i \cdot C^j,$$

The finite-exponential factor A (T) depends on temperature since the transient kinetic energy and the internal degrees of freedom of the molecules influence the probability of the reaction.

The formula determines the rate constant in this case:

$$K = A(T) \cdot \exp(-E/RT),$$

But if A (T) does not depend on temperature, we have Arrhenius's dependence:

$$K = A \cdot \exp(-E/RT).$$

The coefficient of conversion rate K can also be found experimentally by measuring the concentration of formaldehyde at different times of day under different weather conditions.

The results of measurements and the analysis of literary sources regarding the determination of the coefficient of transformation  $K$ , give grounds to assert about its definition with sufficient accuracy for determining the degree of contamination by formaldehyde in places of high concentration of vehicles.

## Conclusions

The approach proposed in this paper allows within the framework of a single model to take into account the complex interconnections between dynamic and physicochemical processes of photochemical smog formation from road transport and to assess the anthropogenic load on the environment, as well as its impact on public health. The model of the formation of formaldehyde as an indicator of the appearance of photochemical smog over transport overpasses in this paper allows to predict secondary photochemical pollution and to make optimal managerial decisions regarding the ecological condition of urbanized territories and to implement environmental measures taking into account the constancy of meteorological factors.

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## **ОЦІНКА ТА ПРОГНОЗ УТВОРЕННЯ ФОТОХІМІЧНОГО СМОГУ НАД ТРАНСПОРТНИМИ ШЛЯХОПРОВОДАМИ В М. КИЄВІ**

**Анотація.** Дана стаття торкається аналізу існуючих моделей формування смогових ситуацій над автомобільними шляхопроводами та в місцях великого скупчення транспорту у великих містах на прикладі м. Києва. Запропоновано математичну модель, яка складається з двох блоків – динамічного та кінетичного, що дозволяє визначати формування теплового купола забруднення та концентрації викидів вуглеводнів в повітрі в залежності від кількості працюючих двигунів. Кінетичний блок моделі дозволяє визначати концентрацію формальдегіду, як індикатора появи фотохімічного смогу в умовах сталості атмосфери. Підраховано концентрацію викидів від автотранспорту в повітрі над основними розв'язками м. Києва при їх повному навантаженні (години пік).

**Ключові слова:** шляхопровід, смог, формальдегід, індикатор забруднення, формування купола забруднення, концентрація забруднень, вуглеводні, швидкість перетворення, математична модель забруднення атмосфери.

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Сіпаков Р.В., Трофімович В.В., Волошкіна О.С., Березницька Ю.О. **Оцінка та прогноз утворення фотохімічного смогу над транспортними шляхопроводами в м. Києві** // Екологічна безпека та природокористування. – 2018. – Вип. 1 (25). – С. 44–51.

Запропоновано математичну модель, що дозволяє визначати формування теплового купола забруднення та концентрації викидів вуглеводнів в повітрі в залежності від кількості працюючих двигунів. Підраховано концентрацію викидів від автотранспорту в повітрі над основними розв'язками м. Києва при їх повному навантаженні (години пік).

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A mathematical model which allows determining the formation of the thermal dome of pollution and the concentration of hydrocarbon emissions in the air, depending on the number of working engines, is proposed. The concentration of emissions from motor vehicles in the air is calculated over the main transport overpasses of the city of Kyiv at their full load (peak hours).

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